

Isolation and identification of stigmasterol from *Byttneria herbacea* Roxb. using methanol extract

M Sathish¹, P K Siva¹, T Parvathi¹, R Bhuvaneswari², M Arumugam^{1*}

¹ PG and Research, Department of Botany, J.J College of Arts and Science (Autonomous), (Affiliated to Bharathidasan University), Pudukkottai, Tamil Nadu, India

² Departments of Physics, Government Arts College for Women (Autonomous), (Affiliated to Bharathidasan University), Pudukkottai, Tamil Nadu, India

Abstract

Phytosterols a group of steroidal alcohol, play an important role in the structure and stability of the cell membrane. The present study deals with characterizing the phytosterol from methanolic extract of *Byttneria herbacea* Roxb. The phytochemical was subjected to physical, chemical, and spectral identification using UV- Vis Spectrophotometer, FTIR, ¹H-NMR, and ¹³C-NMR. Based on the spectral data analysis and chemical reactions, the compounds have been recognized as Stigmasterol.

Keywords: phytosterol, stigmasterol, *Byttneria herbacea*, UV- Vis spectrophotometer

Introduction

The contribution of plants to the life of all living things on earth especially to humans is seen as vital. Various parts of the plant are used as food or medicines without side effects in the treatment of various diseases. Herbal remedies have long been used to treat many ailments (Asila *et al.*, 2018) [1]. Plants are being investigated extensively for their pharmacological potential as a source of raw material for drug discovery. Crude extract of medicinal plants was being used to treat various infectious diseases in traditional medicine systems (Singh *et al.*, 2015 and Hemavathy *et al.*, 2019) [36, 13]. According to the World Health Organization, over 80% of the world's people rely on traditional medicine for their primary healthcare needs (Sasidharan, 2011) [33].

Byttneria herbacea Roxb. Belonging to the Malvaceae family is commonly found in peninsular India (Gujarat, Tamil Nadu, Odisha, and Bihar) (CSIR 1988, Saxena *et al.*, 1994, Tarun *et al.*, 2019) [5, 34, 39]. *B. herbacea* has been described on anti-asthmatic activity (Sharma *et al.*, 2018; Bharathi *et al.*, 2016) [35, 2], anti-inflammatory activity (Sarkar *et al.*, 2013) [31], anti-edematogenic activity (Sarkar *et al.*, 2012) [32], and antioxidant activity (Somkuwar *et al.*, 2014) [37]. Stigmasterol, also known as Stigmasterin or Wulzen anti-stiffness factor is an unsaturated plant sterol present in various plants. It is being used to design synthetic and semi-synthetic compounds in the pharmaceutical industry (Navpreet *et al.*, 2011) [25]. It acts as a forerunner in the synthesis of progesterone and acts as an intermediate in the biosynthesis of androgens, estrogens, corticosteroids, or corticoids (Sundaraman *et al.*, 1977) [38]. It also plays a vital role in the synthesis of vitamin D3 (Kametani *et al.*, 1987) [17].

It has been previously reported in *Physostigma venenosum* (Wind, 1907) [41], *Emilia sonchifolia* (Gao *et al.*, 1993) [8], *Parkia speciosa* (Jamaluddin *et al.*, 1994) [16], *Eclipta alba* (Zhang *et al.*, 1996) [44], *Eclipta prostrata* (Han *et al.*, 1998) [11], *Gypsophila oldhamiana* (Yang *et al.*, 1999) [42], *Croton sublyratus* (De-Eknamkul *et al.*, 2003) [6], *Aralia cordata* (Peng 2005) [29], *Akebia quinata* (Liu *et al.*, 2005) [21],

Heracleum rapula (Luu *et al.*, 2006) [22], *Ficus hirta* (Li *et al.*, 2006) [22, 19], *Eucalyptus globules* (Yang *et al.*, 2007) [43] and *Desmodium styracifolium* (Li *et al.*, 2007) [7, 20]. Since the stigmasterol has been isolated from various plant species, the present study has been made in *B. herbacea* because the species is locally available and grows as a weedy plant.

Materials and Methods

Plant Collection

Byttneria herbacea was collected from the Narthamalai hills, Pudukkottai District, Tamil Nadu, India. The plant material was identified by the Department of Botany, St. Joseph College (Autonomous), Tiruchirappalli, Tamil Nadu, India and the herbarium voucher specimen had been deposited at the PG and Research Department of Botany, J.J. College of Arts and Science, Pudukkottai, Tamil Nadu, India. The fresh plant was collected and dried at room temperature. The dried material was fine powdered by an electric blender and stored in an airtight container.

Extraction

The powdered plant material was extracted with methanol (60-80° C) in the Soxhlet apparatus. The elute was dried and refluxed with water for 8 hrs. The methanol extract was filtered and concentrated using a rotary vacuum evaporator and the dried extract was stored in an airtight container.

Chromatographic Analysis

The methanol extract of *B. herbacea* was subjected to Thin Layer Chromatography (TLC) using silica gel as stationary phase and chloroform: methanol (1:1) as mobile phase. The chromatograms were developed in Iodine chamber. Column chromatography of *B. herbacea* methanol extract (10g) was conducted by wet packing method using silica gel (Mesh 100-200).

The column was run using chloroform, methanol successive by gradient elution technique, and obtained fractions were further studied. A white crystal was formed in any one of the eluates. The crystal was named compound X.

Test for Steroid

The compound X has been tested against Liebermann Buchard and Salkowski reagent for the conformation of the steroidal nucleus (Kandati *et al.*, 2012; Victor and Chidi, 2009) [18, 40].

Spectroscopic Analyses

Spectrophotometric analysis was conducted for the *B. herbacea* methanol extract using a UV-Vis Spectrophotometer. The plant elute was centrifuged for 15 minutes at 3000 rpm and sieved through Whatman No. 1 filter paper. The diluted (1:10) sample was examined under visible and UV lights in wavelengths ranging from 200-1100nm. The characteristic peak values were observed.

FT-IR (Model Shimadzu 8700) has proven to be a valuable tool for the characterization and identification of compounds or functional groups (chemical bonds) present in an unknown sample (Eberhardt *et al.*, 2007 and Hazra *et al.*, 2007) [7, 12]. FT-IR analysis of *B. herbacea* was performed using Perkin an Elmer Spectrophotometer system to denote the characteristic peaks and their functional groups.

The NMR experiment was carried out in BRUKER-AMX400 MHz instrument with 5mg of the purified compound in DMSO_d₆. Tetra Methyl Silane was used as the internal standard and chemical shifts were expressed in ppm.

Result and Discussion

The methanolic extract was yielded 30.62g/kg from aerial parts of *B. herbacea*. The extract was further subjected to column chromatography and verified using TLC in hexane-ethyl acetate 8:2 (v/v) and chloroform-methanol 4:4 (v/v) solvent system. R_f value of the chromatogram was 0.68. The R_f value also corresponds with the previous report on *Acacia nolotica* (Padmasri and Sarada, 2011) [27]. The eluted compounds underwent thin layer chromatography with a slight modification of n-hexane instead of hexane. The isolated compound is confirmed as a steroid by Salkowski and Liebermann-Burchard reaction. Finally, the yield of the isolated compound (figure 5) was 74mg. The isolated compound was powdery with a melting point of 144° - 146° C, and the absorbance of UV- λ_{max}: 255nm (figure. 1 and Table. 1). The molecular formula of the isolated compound also been described by GC-MS data as C₂₉H₃₀O [m/z 413.3(m+)]. FT-IR analysis was performed to identify the functional groups and to identify the formation of hydrogen bonds (Xue *et al.*, 2004; Bao *et al.*, 2006) [30, 4]. Based on the peak values in the region of IR radiation, the functional groups of the compound were separated. The result of FT-IR analysis confirmed the presence of functional groups at the

peak value 3430.14 (N-H stretching primary amine), 2922.07 (C-H stretching alkane), 2852.67 (C-H stretching alkane), 2086.39 (N=C=S stretching isothiocyanate), 1642.63 (C=C stretching alkene), 1469.84 (C-H bending alkane methylene group), 1378.52 (S=O stretching sulfonyl chloride), 1249.75 (C-O stretching alkyl aryl ether), 1215.65 (C-O stretching vinyl ether), 1113.16C-O stretching secondary alcohol 1070.96 (S=O stretching sulfoxide), 752.82 (C=C bending alkene) (Figure 2 and Table 1). These absorption frequencies resembled the previous reports of Stigmasterol by Muthukrishnan *et al.*, (2015) and Olawumi (2019) [24, 26].

The proton NMR (Figure 3 and Table 1) showed the proton of H-3 appeared as a multiplet at 3.51ppm and this reveals the existence of signal for olefinic proton at 4.98, 5.14, and 5.31. The angular methyl protons showed at 0.71, 0.80, and 1.02 corresponds to C19, C20, and C21 protons respectively (Habib *et al.*, 2007) [9, 10].

¹³C-NMR supporting evidence for the structure of the Stigmasterol compound was provided by the analysis of ¹³C-NMR data and a complete assignment is given below (Figure 4 and Table 1). Due to hydroxylation at 3-position, C-3 carbons absorb at δ 72.12ppm. The ¹³C has recognizable signals at 141.15ppm and 121.85ppm which are assigned C5 and C6 double bonds respectively. The value at 21.62, 138.72, and 129.60ppm corresponds to angular carbon C19, C20 and C21. Spectra show twenty-nine carbon signals including six methyls, nine methylene, eleven methane, and three quaternary carbon (Jamal *et al.*, 2019) [14, 15].

Conclusion

The phytochemical assessment of the methanol fraction of the aerial parts of *B. herbacea* belonging to the family Malvaceae was efficiently carried out. From these chemical and spectral pieces of evidence, compound X was confirmed as Stigmasterol (Figure 5). Now, it is the turn of the pharmacologists/scientists to research the plant to explore the content by individual bioactivity of the stigmasterol. So, the current study will upgrade scientific networks to accomplish more work on this significant therapeutic plant in not so distant future.

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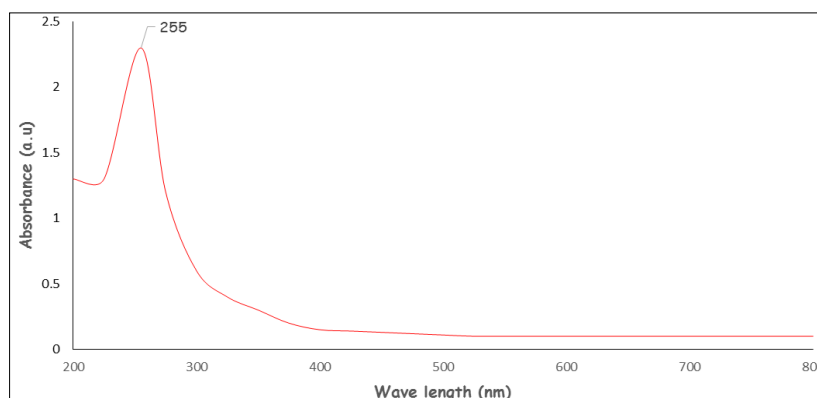


Fig 1: UV-visible spectrophotometer

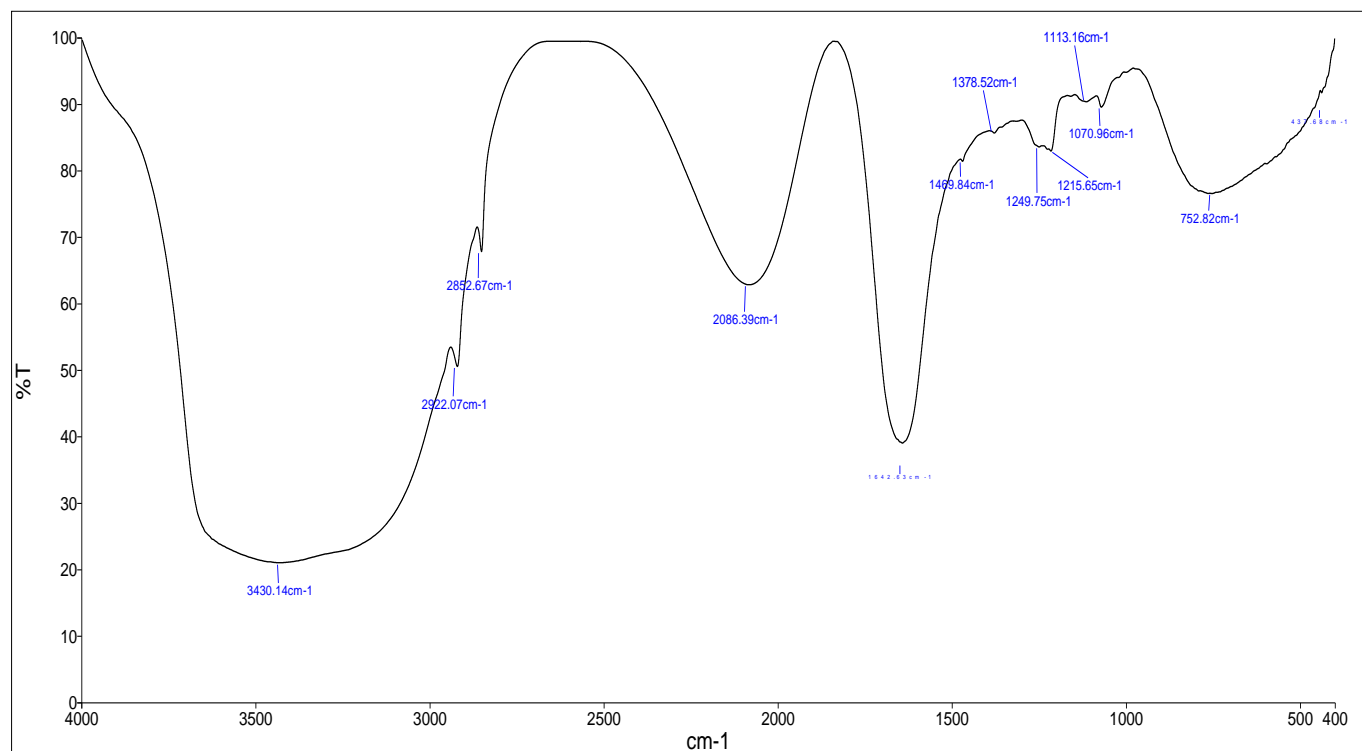


Fig 2: Fourier transform infrared spectrum (FTIR) Analysis

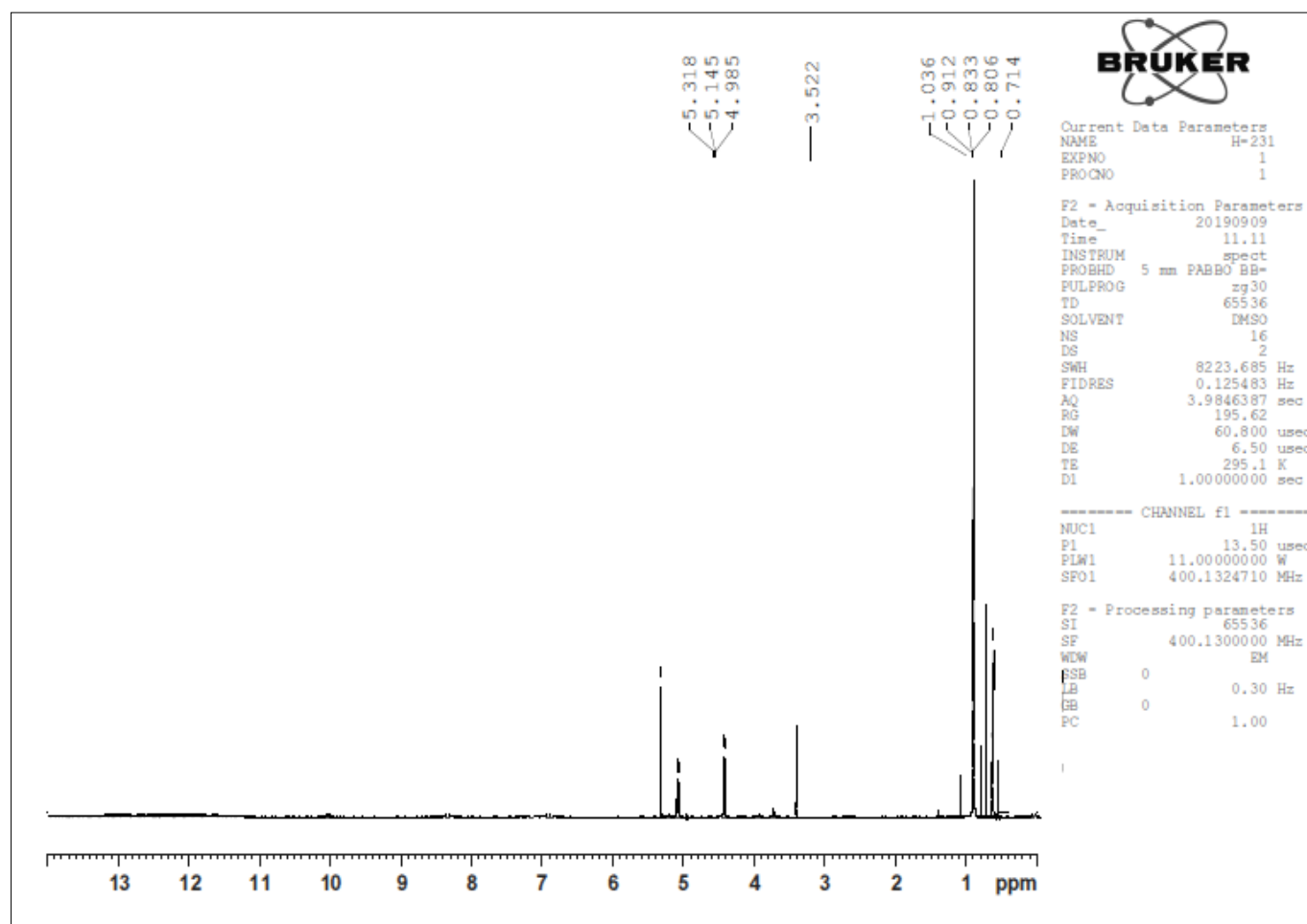
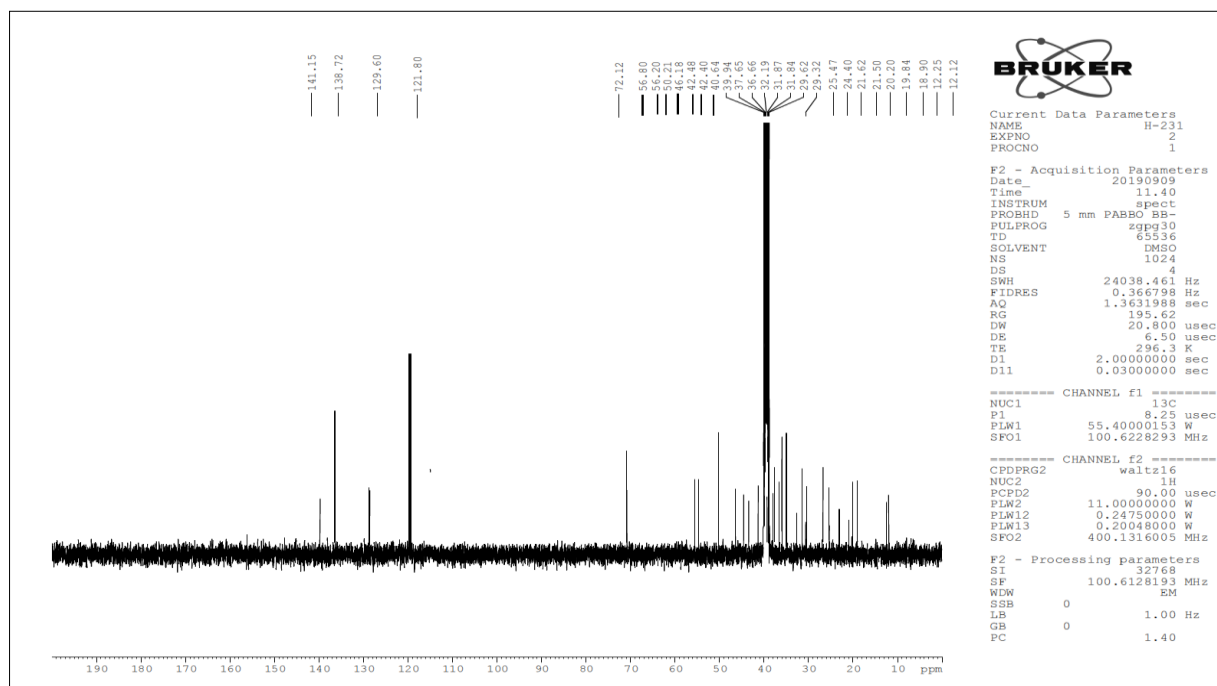


Fig 3: ¹H-NMR Spectrum

Fig 4: ¹³C-NMR SpectrumTable 1: Spectroscopic Data of Isolated Compound from Methanol Fraction of *B. herbacea*

Spectroscopic techniques	Data
UV λ max	255nm
IR (ranges in cm-1)	3430.14 (N-H stretching primary amine), 2922.07 (C-H stretching alkane), 2852.67 (C-H stretching alkane), 2086.39 (N=C=S stretching isothiocyanate), 1642.63 (C=C stretching alkene), 1469.84 (C-H bending alkane methylene group), 1378.52 (S=O stretching sulfonyl chloride), 1249.75 (C-O stretching alkyl aryl ether), 1215.65 (C-O stretching vinyl ether), 1113.16 (C-O stretching secondary alcohol), 1070.96 (S=O stretching sulfoxide), 752.82 (C=C bending alkene) functional groups
¹ H NMR (DMSO)	δ 0.71, δ 0.80, δ 0.82, δ 0.83, δ 0.91, δ 1.03 (each 3H, Me × 6), δ 3.51 (¹ H, m, H-3), δ 5.31 (1H, t, H-6), δ 5.14 (¹ H, s, H-22), δ 4.98 (¹ H, s, H-23) ppm.
¹³ C NMR (DMSO)	δ 37.65, δ 32.19, δ 72.12, δ 42.48, δ 141.15, δ 121.85, δ 31.84, δ 31.87, δ 50.21, δ 36.66, δ 21.50, δ 39.94, δ 42.40, δ 56.80, δ 24.40, δ 29.32, δ 56.20, δ 40.64, δ 21.62, δ 138.72, δ 129.60, δ 46.18, δ 25.47, δ 12.12, δ 29.62, δ 20.20, δ 19.84, δ 18.90, δ 12.25.

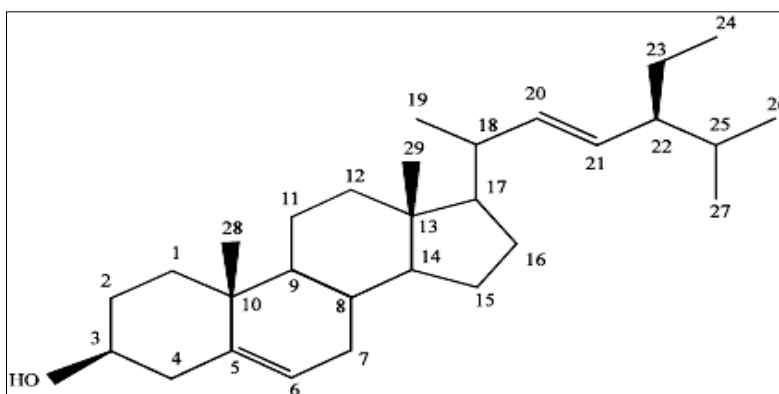


Fig 5: Structure of Stigmasterol

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