

## Estimation of total phenolic and flavonoid contents and evaluation of antioxidant activity of different parts of *Musa acuminata* cv. grand nain and *Musa balbisiana*

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### Abstract

Plants are known as good sources of natural therapeutic agents. Plants generally contain secondary metabolites like phenolics, flavonoids, glycosides, coumarins, saponins, terpenoids and alkaloids *etc.* which reveal their specific characteristic properties and attribute to their pharmacological properties. Bananas are popularly consumed fruits in world because of its simple taste, high nutritional value and potential health benefits. In the present investigation, different parts of two species of banana *Musa acuminata* cv. Grand Nain (flowers, root, bract, leaves) and *Musa balbisiana* (flowers, root, bract, leaves, seeds) were subjected to estimate total phenolic and flavonoid contents as well as extracts of these plant parts were also evaluated for their antioxidant activity by DPPH assay and peroxidase (POX) enzyme assay. Results of the present study indicated the presence of good quantity of total phenols and flavonoid contents in all the selected plant parts. The maximum phenolic content in *Musa acuminata* cv. Grand Nain was found in leaves ( $28.985 \pm 5.65$  mg GAE/gdw) compared to flowers, Bract, and roots while flowers of the plant were found to possess the highest flavonoid content ( $64.87 \pm 9.62$  mg QE/gdw) than other parts. In *Musa balbisiana*, flowers were found to have the maximum phenolic content ( $28.96 \pm 4.26$  mgGAE/gdw) while, total flavonoid contents were found in leaves of the plant. The maximum free radical scavenging activity in *Musa acuminata* cv. Grand Nain was shown by roots with the lowest  $IC_{50}$  value ( $234.7 \pm 10.25$  mg/L) while by bract in *Musa balbisiana* ( $234.65 \pm 12.58$  mg/L). The maximum peroxidase activity was shown by acetone extract of roots of *Musa acuminata* cv. Grand Nain ( $0.412 \pm 0.034$  mMole/min/g.dw). All experiments were done in triplicates. One way ANOVA analysis showed lack of significant differences in antioxidant activity and total phenolic and flavonoid contents among all the selected plant parts.

**Keywords:** *Musa acuminata* cv. grand nain, *Musa balbisiana*, total phenols, total flavonoids, antioxidant activity etc

### Introduction

Various plants have been used for therapeutic purpose in different countries since ancient and are a good source of potent and powerful drugs. Since time immemorial people have tried to find medications to alleviate pain and cure different illnesses [1]. Plants generally contain secondary metabolites like phenolics, flavonoids, glycosides, coumarins, saponins, terpenoids and alkaloids *etc.* which reveal their specific characteristic properties and attribute to their pharmacological properties [2]. The phytoconstituents have a major potential for developing phytomedicines which are considered to be generally safe.

As shown in recent years, natural antioxidants discovered in plants have attracted some interest due to their widely acclaimed nutritional and therapeutic values. Antioxidant properties stand to be an essential mechanism of beneficial activity of plant-derived compounds and extracts. Ethnopharmacological surveys have shed light on the fact that the therapeutic use of even 80% of 122 plant derived drugs may have a link with their recommendations in traditional medicine [3]. Natural antioxidants have a diversity of biochemical activities, some of which include the inhibition of reactive oxygen species (ROS) generation, direct or indirect scavenging of free radicals, and alteration of intracellular redox potential [4]. Antioxidants have functioned to inhibit apoptosis because apoptosis was at first thought to be mediated by oxidative stress [5]. It is known that many antioxidant substances have anticancer or anticarcinogenic properties [6,7]. Epigallocatechin-3-gallate

(EGCG) in green tea, for instance, has been reported to scavenge free radicals [8] and to hinder carcinogen-induced tumours in the skin, lung, forestomach, and colon of rodents [9]. Therefore, there has been undeniable evidence of interest when it comes to finding natural antioxidants from plant materials.

Studies regarding the bioactivities of various plants have assumed an important position because of the variations in the effectiveness of the plant extract with the solvent for extraction used, plant part used, the plants' age, and geographic origin. The excessive use of medicinal plants for drug formulation also puts pressure on the need for more biomass of plants which can be met with biotechnological tools like micropropagation.

Grand Nain bananas are cultivars of *Musa acuminata*. It is the most commonly cultivated banana. Nearly all banana cultivars are descendants and/or hybrids of *Musa acuminata* or *Musa balbisiana*. *Musa balbisiana*, also known as plantain, is a wild type banana species. It is one of the ancestors of modern cultivated bananas, along with *Musa acuminata*. Bananas are popularly consumed fruits in world because of its simple taste, high nutritional value and potential health benefits [10].

In the present study, antioxidant activity along with total phenolic and flavonoid contents was evaluated in different parts of two species of banana i.e. *Musa acuminata* cv. Grand Nain (flowers, root, bract, leaves) and *Musa balbisiana* (flowers, root, bract, leaves, seeds) which belong to family Musaceae.

## Materials and Methods

### Sample Collection and Processing

Plant samples, *Musa acuminata* cv. Grand Nain (flowers, root, bract, leaves) were collected from Surat, Gujrat, India and *Musa balbisiana* (flowers, root, bract, leaves, seeds) were collected from University of Rajasthan, Jaipur, Rajasthan, India at different times. All samples were free from microbial and physical damage. The samples were washed, and the plant parts were separated. Those were shade dried at room temperature for 10 days. The dried parts were grounded in powdered form. The powdered sample were stored in an airtight container separately and were kept to be used for further phytochemical analysis.

### Sample Extraction

The dried powdered of plant parts were extracted by cold percolation method using methanol as a solvent. 10 g of the dried powder was taken in a conical flask having 100 ml methanol and kept in an orbital shaker at 120 rpm for 24 h. After 24 h, the extracts were filtered through Whatman filter paper no.1 for removal of peel particles and evaporated under vacuum.

### Determination of Total Phenolic Contents in the Plant Extracts

TPC (The total phenolic content) was determined by the Folin-Ciocalteu method<sup>[11,12]</sup>. The reaction mixture was prepared by mixing 0.5 ml of methanolic solution (1 mg/ml) of extract, 2.5 ml of 10% Folin-Ciocalteu's reagent dissolved in water, and 2.5 ml 7.5% NaHCO<sub>3</sub>. The mixture was allowed to stand for 15 min at 45°C, and the phenols were determined by the spectrophotometric method. The absorbance was determined at  $\lambda_{\text{max}} = 765 \text{ nm}$ . The samples were prepared in triplicate, and the mean value of absorbance was obtained. Blank was concomitantly prepared, with methanol instead of extract solution. The standard curve was prepared using the standard solution of Gallic acid in methanol in the range 100-1000 mg/L. The total phenolic content was expressed in terms of gallic acid equivalent (mg of GAE/g of dry weight), which is a common reference compound.

### Determination of Total Flavonoid Concentrations in the Plant Extracts

The concentration of TFC (total flavonoid content) was determined using aluminium chloride spectrophotometric method<sup>[13]</sup> with slight modifications. Plant extracts (0.5 ml) were dissolved with 1.5 ml of methanol, 0.1 ml of 10% aluminium chloride, 0.1 ml of 1 M potassium acetate, and 2.8 ml of distilled water and incubated for half an hour at room temperature. The absorbance of the reaction mixture was measured at 415 nm. All experiments were prepared in triplicate, and the mean value of absorbance was obtained, and values were expressed in mean  $\pm$  standard deviation. The standard curve was prepared using the standard solution of quercetin in methanol. Total flavonoid content of the extracts was expressed in milligrams of quercetin equivalents per gram dry weight.

### Extraction of flavonoids

Flavonoids were extracted from different parts of theselected plants following the well-established method<sup>[14]</sup>. 100 grams of finely powdered plant parts were Soxhlet extracted with hot 80% methanol (500ml) on a water bath

for 24 h and filtered. Filtrate wasre-extracted successively with petroleum ether, ethylether and ethyl acetate. Each step was carried out three times to ensure complete extraction. Petroleum ether fraction was discarded due to being rich in fatty substances and ethyl ether fractions (free flavonoids) were collected. Ethyl acetate fractions were analysed for bound flavonoids. Each fraction was hydrolysed in 7% H<sub>2</sub>SO<sub>4</sub> for 2 h. Resulting mixture was filtered and filtrate was again extracted with ethyl acetate. The ethylacetate extract was washed with distilled water till neutrality and collected. The ethyl ether(free flavonoids) and ethyl acetate fractions (boundflavonoids) were dried *invacuo* weighed and storedin glass vials at 40° C till used.

### Determination of Antioxidant Activity

#### DPPH Radical Scavenging Activity

The DPPH radical scavenging activity of the extracts were evaluated by 1, 1 - diphenyl 2 - picryl - hydrazil (DPPH) using the method given by Bhat and Karim<sup>[15]</sup>. An aliquot (100 $\mu$ l) of peel extract was mixed with 3.9 ml of 0.1 mM DPPH methanolic solution. The mixture was vortexed thoroughly and kept in the dark for 30 min. The absorbance was measured at 515 nm, against a blank of methanol. The radical's scavenging activity was calculated using;  $(\text{Ab}_{\text{control}} - \text{Ab}_{\text{sample}} / \text{A}_{\text{control}}) \times 100$  Where,  $\text{Ab}_{\text{control}}$  is the absorption of the DPPH solution and  $\text{Ab}_{\text{sample}}$  is the absorption of the DPPH solution after the addition of the sample. A linear graph of concentration vs percentage inhibition was prepared, and IC<sub>50</sub> values were calculated. The antioxidant activity of each sample was expressed in terms of IC<sub>50</sub> (defined as the amount of concentration required to inhibit DPPH radical formation by 50%), calculated from the inhibition curve. Ascorbic acid and BHT (Butylated Hydroxy Toluene) were used as a reference.

#### Peroxidase assay (POX)

200 mg Plant sample was homogenised with 10 ml of phosphate buffer and refrigerated and centrifuged at 10000 rpm for 20 min. the clear supernatant was taken as the enzyme extract. The activity was assayed after the method of Chance & Maehly<sup>[16]</sup> with the following modifications. 2.4 ml of phosphate buffer, 0.3 ml of pyrogallol and 0.2 ml of water was added. The amount of purpurogallin formed was determined by taking the absorbance at 420 nm immediately after adding 0.1 ml enzyme extract. The extinction coefficient of 2.8 mM-1cm-1 was used in calculating the enzyme activity that was expressed in terms of mMole per minute per gram dry weight.

Peroxidase value = O.D. value  $\times$  2.8 mM<sup>-1</sup>cm<sup>-1</sup>

#### Statistical Analysis

All experimental results were carried out in triplicate and were expressed as the average of three analyses with Standard Deviation. The IC<sub>50</sub> values were also calculated by linear regression analysis.

### Results and Discussion

Plants contain many phytochemicals which are useful sources of natural antioxidants such as phenols, flavonoids, tannins, phenolic acids etc<sup>[17]</sup>. Various studies have been reported that presence of good phenolic and flavonoid contents in plants are indicator of their antioxidant capacity. Flavonoids are a widely distributed group of phenols that act as effective antioxidants. Being plant secondary metabolites,

the phenolics or polyphenols are very important judging from the virtue of their antioxidant activities by chelating redox-active metal ions, inactivating lipid free radical chains, and avoiding the hydroperoxide conversions into reactive oxyradicals [18]. Results of the present study revealed that both the selected species (*Musa acuminata* cv. Grand Nain and *Musa balbisiana*) are good sources of phenolic and flavonoid contents (Table 1-2, Figure 1). The maximum phenolic content in *Musa acuminata* cv. Grand Nain was found in leaves ( $28.98 \pm 5.65$  mg GAE/gdw) compared to flowers, Bract, and roots while flowers of the plant were found to possess the highest flavonoid content ( $64.87 \pm 9.62$  mg QE/gdw) than other parts. In *Musa balbisiana*, flowers were found to have the maximum phenolic content ( $28.96 \pm 4.26$  mgGAE/gdw) while, total flavonoid contents were found in leaves of the plant. Free and bound flavonoids were also extracted from plants and from all plant parts, good amount of both types of flavonoids were isolated (Figure 2). Result of one-way ANOVA showed lack of significant differences in phenolic and flavonoid contents in various plant parts and between different species. Presence of good amount of phenolic and flavonoid contents showed that both the species are good sources of natural antioxidants as the rich-flavonoid plants could manifest themselves as good sources of antioxidants that would assist in the enhancement of the overall antioxidant capacity of an organism and protection against lipid peroxidation [19]. The polyphenol anti-oxidant capacity has been taken into account as one of the outstanding mechanisms of action in inhibiting mutagenesis and cancer initiation by means of their capacity to scavenge ROS, activate antioxidant enzymes, prevent carcinogen-induced DNA adduct formation, enhance DNA repair and reduce overall oxidative DNA injury [20]. DPPH is commonly used assay to evaluate free radical scavenging activity of plant

extract. In the present investigation, ethanolic extracts of the selected plant parts were subjected to DPPH assay at different concentrations (0-100 mg/L) and results showed that the selected parts of both species have great potential to scavenge free radicals (Table 3, Figure 3). The maximum free radical scavenging activity in *Musa acuminata* cv. Grand Nain was shown by roots with the lowest IC<sub>50</sub> value ( $234.7 \pm 10.25$  mg/L) while by bract in *Musa balbisiana* ( $234.65 \pm 12.58$  mg/L). IC<sub>50</sub> value by standard compound was found to be  $76.22 \pm 3.62$ . No significant difference was found in IC<sub>50</sub> values of different parts, and in different species by one way ANOVA analysis.

While hydrogen peroxide itself is not very reactive, it can sometimes be poisonous to cells, since it may trigger the rise of hydroxyl radicals inside the cell [21]. Peroxidase enzymes are known as enzymatic antioxidants which help to scavenge peroxides. In the present investigation, ethanolic and acetone extracts of different parts of *Musa acuminata* cv. Grand Nain and *Musa balbisiana* were subjected to peroxidase assay (POX) and results showed presence of good peroxidase activity by all the extracts of both species (Table 4, Figure 4). The maximum peroxidase activity was shown by acetone extract of roots of *Musa acuminata* cv. Grand Nain ( $0.412 \pm 0.034$  mMole/min/g.dwt).

In previous studies, it has also been shown that Chemical compounds, existing with different properties in many plants, are spread throughout the plant's organs. Banana peel contains flavonoid compounds whose properties include the potential for antioxidants [22]. It also contains many carbohydrates, minerals such as potassium and sodium, and cellulose. Based on a phytochemical analysis of banana peel extract [23] affirm that banana peels contain secondary metabolites, such as saponins, tannins, alkaloids, flavonoids, phlobatannins, anthraquinones, and quinones, that have antibacterial activity [24].

**Table 1:** Total phenolic, total flavonoids, and isolated flavonoid contents in *Musa acuminata* cv. Grand Nain.

Plant part	Total phenolic content (mg GAE/gdw)	Total flavonoid content (mg QE/gdw)	Isolated flavonoid (mg/d.dwt)		
			Free (F)	Bound (B)	Total (F+B)
Leaves	$28.98 \pm 5.65$	$58.86 \pm 10.87$	2.155	1.630	3.785
Flowers	$22.47 \pm 9.84$	$64.87 \pm 9.62$	2.488	0.754	3.242
Bract	$15.82 \pm 4.28$	$16.24 \pm 2.54$	1.984	0.941	2.925
Root	$1.25 \pm 0.08$	$8.59 \pm 1.58$	1.254	1.591	2.845

**Table 2:** Total phenolic, total flavonoids, and isolated flavonoid contents in *Musa balbisiana*.

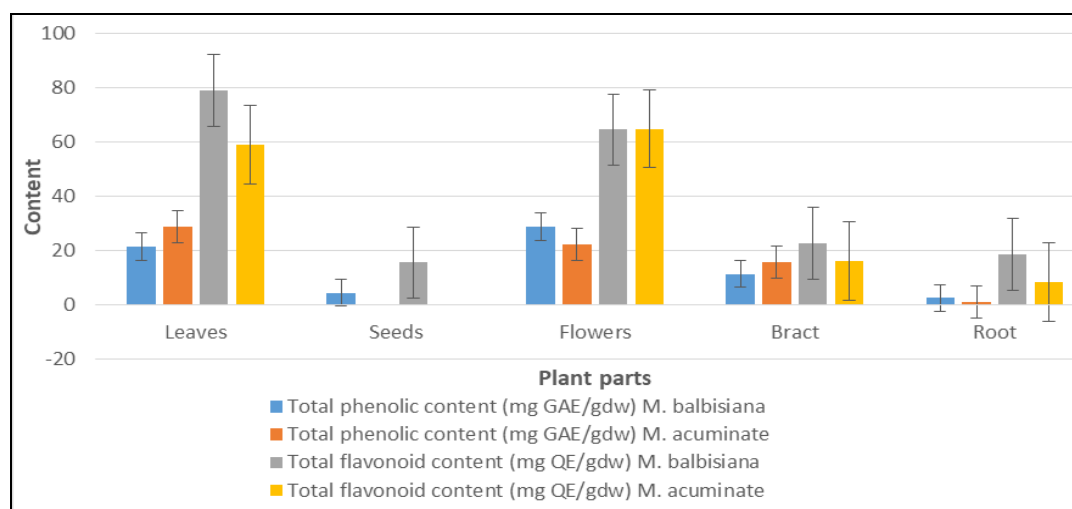
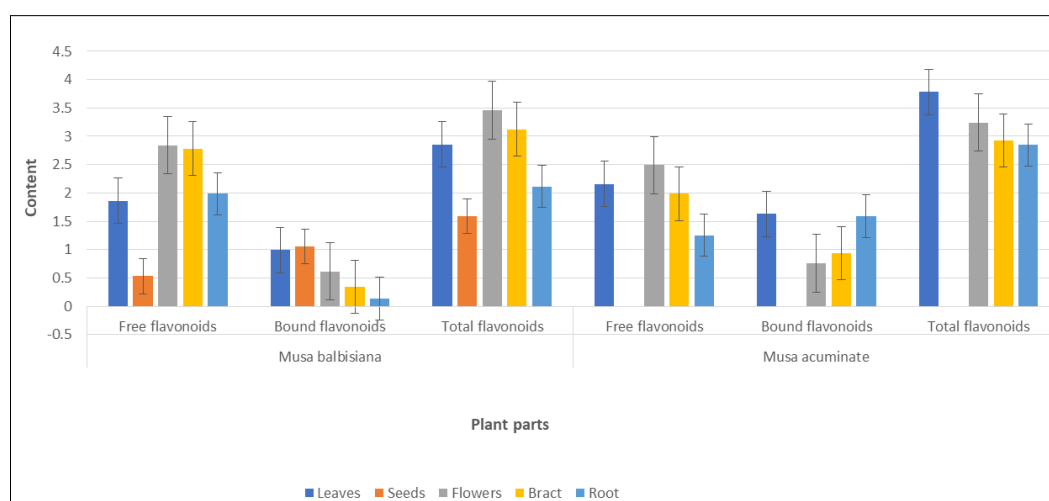
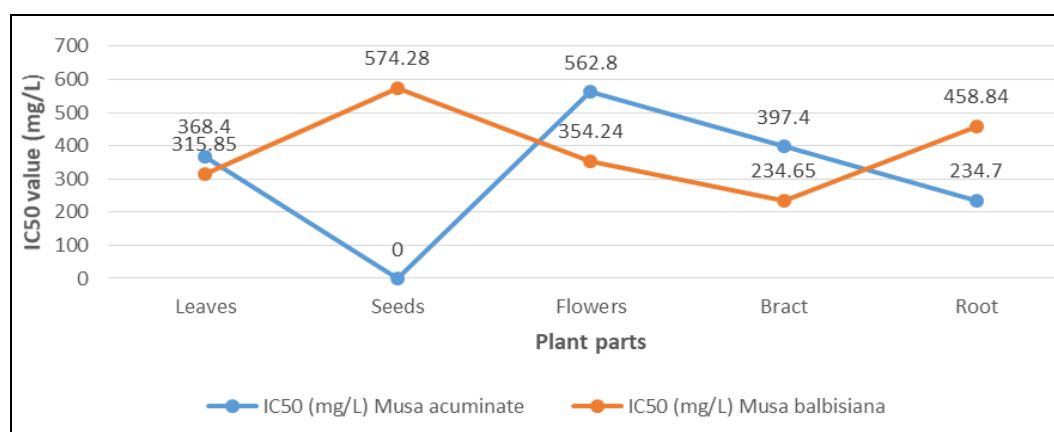
Plant part	Total phenolic content (mg GAE/gdw)	Total flavonoid content (mg QE/gdw)	Isolated flavonoids (mg/g.dwt)		
			Free (F)	Bound (B)	Total (F+B)
Leaves	$21.58 \pm 3.72$	$78.93 \pm 10.35$	1.858	0.994	2.852
Seeds	$4.58 \pm 1.15$	$15.64 \pm 1.58$	0.528	1.059	1.587
Flowers	$28.96 \pm 4.26$	$64.52 \pm 12.87$	2.842	0.616	3.458
Bract	$11.52 \pm 1.63$	$22.85 \pm 5.97$	2.784	0.341	3.125
Root	$2.58 \pm 0.56$	$18.54 \pm 2.52$	1.985	0.13	2.115

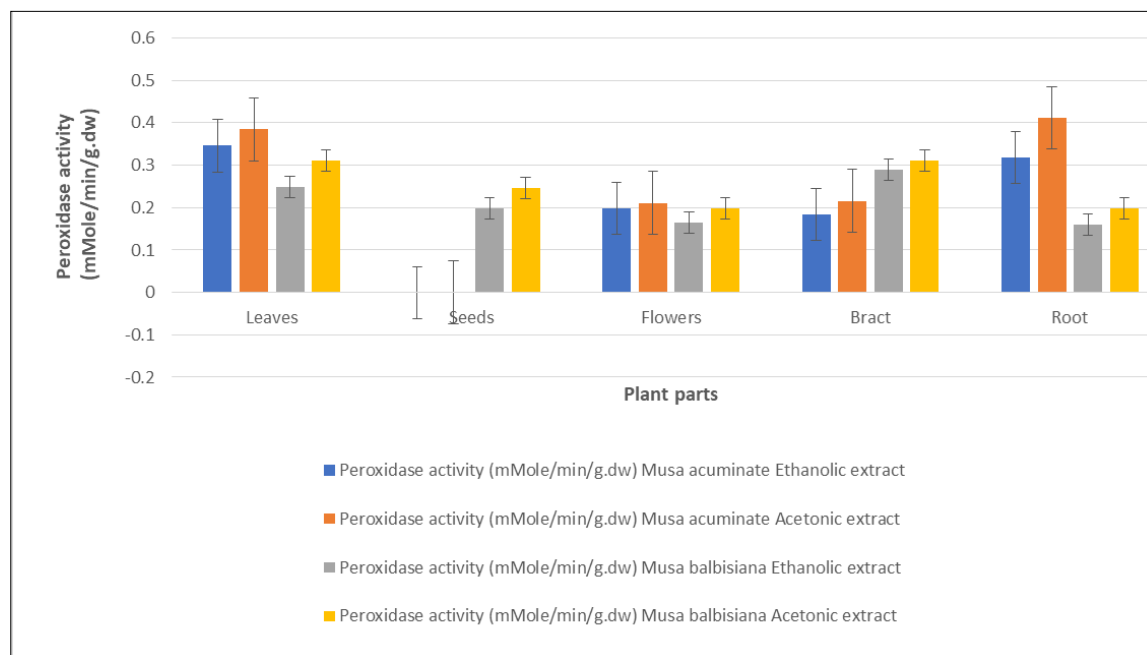
**Table 3:** Evaluation of antioxidant activity of plant parts by DPPH assay.

Name of plant part	IC <sub>50</sub> (mg/L)	
	<i>Musa acuminata</i> cv. Grand Nain	<i>Musa balbisiana</i>
Leaves	$368.4 \pm 15.85$	$315.85 \pm 14.58$
Seeds	-	$574.28 \pm 12.82$
Flowers	$562.8 \pm 12.54$	$354.24 \pm 14.35$
Bract	$397.4 \pm 18.15$	$234.65 \pm 12.58$
Root	$234.7 \pm 10.25$	$458.84 \pm 18.87$

**Table 4:** Evaluation of antioxidant activity by peroxidase (POX) enzyme assay.

Name of plant part	Peroxidase activity (mMole/min/g.dw)			
	<i>Musa acuminata</i> cv. Grand Nain		<i>Musa balbisiana</i>	
	Ethanollic extract	Acetonic extract	Ethanollic extract	Acetonic extract
Leaves	0.346±0.012	0.385±0.152	0.248±0.012	0.312±0.021
Seeds	-	-	0.198±0.120	0.246±0.026
Flowers	0.198±0.054	0.211±0.024	0.165±0.023	0.198±0.010
Bract	0.183±0.024	0.216±0.058	0.289±0.029	0.310±0.183
Root	0.319±0.120	0.412±0.034	0.159±0.034	0.198±0.158

**Fig 1:** Total phenolic and flavonoid contents in different parts of two species of banana.**Fig 2:** isolated flavonoid contents from different parts of two species of banana.**Fig 3:** Evaluation of antioxiant activity of different parts of two species of banana by DPPH assay.



**Fig 4:** Evaluation of peroxidase enzyme activity in different extracts of two species of banana.

### Conclusion

The present work shows the presence of high flavonoid and phenolic contents in different parts of two species of banana (*Musa acuminata* cv. Grand Nain and *Musa balbisiana*) which is common edible plant. Results of the study further suggested that these plant parts are rich source of natural antioxidants and eating banana provide good antioxidants and can protect us from various diseases. The present study may help researchers and scientists to make strategies of developing antioxidant rich varieties of food crops.

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